

Spherisorb 5 μ m ODS. Elution was isocratic with either H₂O–MeOH (9:1) or H₂O–MeOH (7:3) containing 10^{−3} N HCl, flow rate 1 ml min^{−1} at 155 kg cm².

GLC. Carried out on ABA Me ester with an instrument equipped with an electron-capture detector and a 130 \times 0.2 cm glass column packed with 3% DC-200 on Gas-Chrom Q. The carrier was Ar–CH₄ (19:1), flow rate 40 ml min^{−1}, oven temp. 165° and detector temp. 186°.

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NORBOTRYAL ACETATE, A NOR-SESQUITERPENOID ALDEHYDE FROM *BOTRYTIS CINEREA*

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Key Word Index—*Botrytis cinerea*; Hyphomycetes; sesquiterpenoid; nor-sesquiterpenoid; botrydial; ¹³C NMR.

In the course of biosynthetic studies [1] on the metabolites of the plant pathogen, *Botrytis cinerea*, we isolated an unstable oily acetoxy-aldehyde, C₁₆H₂₄O₃. The IR [ν_{\max} 2740, 1670, 1625 (C=C·CHO) and 1735, 1235 (CH₃·CO·O) cm^{−1}] and NMR [δ 10.0 (1H, s) and 2.00 (3H, s)] revealed the presence of the aldehyde and acetate groupings. Oxidation with CrO₃ and methylation of the product with CH₂N₂ gave a monomethyl ester which was again an unstable oil. The NMR data (see Table 1) can be interpreted in terms of structure (1) for the compound which we have named norbotryal acetate. The fully substituted nature of the $\alpha\beta$ -unsaturated aldehyde was confirmed by the UV [λ_{\max} 253 nm (calc. 254 nm), ϵ 10300] and by the singlet character of the olefinic ¹³C resonances. PMR decoupling experiments established the presence of a CH₂·CH(OAc)·CH— and two CH·Me groups. The location of the other carbon atoms in two tertiary methyl groups, a methylene, a quaternary carbon atom and an acetoxy group, was clear from the ¹H and ¹³C NMR spectra.

The co-occurrence of this compound with botrydial (2) [2] led to the proposed structure (1). This relationship

Table 1. The NMR spectra of norbotryal acetate

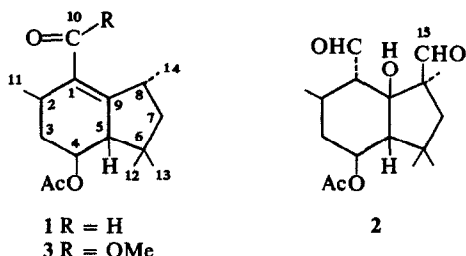
Position	¹³ C resonance*	¹ H resonance
1	135.8	
2	32.5	2.85 (m)
3	37.5	1.38 and 2.14 (m)
4	70.2	4.75 (octet $J = 3, 9, 10$ Hz)
5	56.0	2.54 (q $J = 2$ and 9 Hz)†
6	40.9	
7	49.3	1.8 (m)
8	29.1	3.2 (m)†
9	169.0	
10	190.4	10.0 (s)
11	21.9	1.12 (d, $J = 6$ Hz)
12	28.3	1.25 (s)
13	20.6	0.75 (s)
14	24.2	1.28 (d, $J = 7$ Hz)
C=O	170.1	
Me	21.3	2.00 (s)

(determined in CDCl₃; values in ppm from TMS)

* Assigned by comparison with botrydial and derivatives.

† These signals showed a long-range coupling.

was confirmed by heating botrydial with alumina which gave a low yield of norbotryal acetate through dehydration and subsequent deformylation of the vinylogous β -dialdehyde. The β -orientation of the hydrogen at C-8 was suggested by the long-range 'W' coupling (2 Hz) between the C-5 and C-8 protons which was established by decoupling experiments. This interrelationship raised



the possibility that norbotryal acetate was an artefact of the purification procedure. However it may be detected TLC and by GLC of a crude EtOAc broth extract. This does not of course preclude its formation by non-enzymatic means from botrydial or a C-15 carboxylic acid, in the culture broth whilst the fungus was growing.

EXPERIMENTAL

Isolation. *Botrytis cinerea* (GCRI strain 216) was grown for 10 days on surface culture on a Czapek Dox medium (101.) containing 0.1% yeast extract and 5% glucose. The broth was extracted with EtOAc and separated into acidic and neutral fractions with aq. NaHCO₃. The neutral fraction (4.0 g) was chromatographed on Si gel (250 g) (Merck, deactivated with

12% H₂O). Elution with 15% EtOAc-petrol gave a fraction which was further purified by PLC on Si gel in EtOAc-petrol (1:4) to afford norbotryal acetate (1) (476 mg) as an unstable oil. The material was homogeneous by TLC; in the above system it had R_f 0.53 and gave a pink colouration with a MeOH-H₂SO₄ spray. It was also homogeneous by GLC on 1% OV17 at 170°. It had $[\alpha]_D^{20} + 102^\circ$ (c 0.2 in CHCl₃); bp 84°/0.5 mm; MS 264, 204 (base peak), 189, metastable 175.1 (for 204-189), 161, 119, 105, 43, 41. Accurate mass: found 204.151481, C₁₄H₂₀O⁺ (M - MeCO₂H) requires 204.151407.

Oxidation. The aldehyde (55 mg) in Me₂CO (5 ml) was treated with the 8N CrO₃ reagent (0.5 ml) for 1 hr. MeOH was added, the soln conc in *vacuo* and the product recovered in EtOAc. It was methylated with CH₂N₂ in Et₂O and purified by PLC to afford the methyl ester (3) as an oil, ν_{\max} 1730, 1680, 1630 cm⁻¹, δ 0.80 (3H, d, J = 7 Hz), 0.95 (3H, d, J = 6 Hz), 1.05 (3H, s), 1.25 (3H, s), 2.00 (3H, s), 3.75 (3H, s), 4.8 (1H, m). MS 294, 250, 235, 208, 203, 193, 175, 43.

Reaction of botrydial with alumina. Botrydial (50 mg) in dry C₆H₆ (10 ml) was heated at 60° with Al₂O₃ (1 g) (dried in a vacuum oven at 120° overnight) for 1 hr. The soln was filtered, washed with aq. NaHCO₃, dried and evaporated. The residual gum was purified by PLC on Si gel in EtOAc-petrol (1:4) to afford norbotryal acetate (4 mg) identified by its IR spectrum.

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HALOGENATED SESQUITERPENES FROM THE MARINE RED ALGA *MARGINISPORUM ABERRANS*

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Key Word Index—*Marginisporum aberrans*; Corallinaceae; red alga; sesquiterpenes; laurinterol; isolaurinterol; aplysin; aplysinol; aplysinal.

In our previous paper [1], *p*-hydroxybenzaldehyde, dichloroacetamide and 3,5-dinitroguaiacol were identified as antimicrobial components of the marine red alga *Marginisporum aberrans*. Further study of the *n*-hexane extracts of the same plant has resulted in the isolation of five bromine-containing sesquiterpenes, which are described in the present communication.

The *n*-hexane-soluble fraction of the MeOH extracts of the seaweed (15 kg) was chromatographed on Si gel. From fractions eluted with *n*-hexane-Et₂O (3:1), there were obtained 10 mg laurinterol (1) [2-5] and a mixture of bromo-compounds (12 mg) from which isolaurinterol (2) [2], aplysin (3) [2,3,5,6], aplysinol (4) [3,6] and a new compound (5) were isolated by preparative GLC (2 m ×